

260(a). Solution for \underline{V}

The starting equation is:

$$(V-E)^3 - a(V-E) + b = 0 \quad - (1)$$

where

$$a = \frac{\hbar^2}{2m} \nabla^2 V \quad - (2)$$

$$b = \frac{\hbar^2}{2m} \underline{\nabla} V \cdot \underline{\nabla} V \quad - (3)$$

Eq. (1) is a cubic in $V-E$. It can be written as:

$$E^3 - 3VE^2 + (3V^2 + a)E + V^3 - aV + b = 0 \quad - (4)$$

which is a cubic for E .

Using computer algebra the three roots of the cubic (1) can be found, giving three solutions for $V-E$. Alternatively the three roots of eq. (4) can be found for E . The algebra of eq. (4) can be checked by computer therefore E can be found as a function of V , $\underline{\nabla} V$ and $\nabla^2 V$.

Now use the fact that the total energy E is constant, so:

$$\underline{\nabla} E = 0 \quad - (5)$$

for each of the three roots of eq. (1).

2) Eq. (5) gives a differential equation in V ,
 an equation that can be solved numerically.
 Finally we use this V in the Schrodinger equation:

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + V\psi = E\psi \quad - (6)$$

and find the wave functions ψ and energy levels
 E of the particles of any elementary particle.

Analytical Solution of Eq. (1)
 The solution of the cubic:

$$x^3 + px = q \quad - (7)$$

can be found using:

$$x = w - \frac{p}{3w} \quad - (8)$$

Then
$$w^3 = \frac{1}{2} \left(q \pm \left(q^2 + \frac{4}{27} p^3 \right)^{1/2} \right)$$