

306(6) General Rule for Number of Lines in the series
 $n' = N$ to $n = N + 1$, Left Circular Polarization

To establish this line we consider:

- 1) $n' = 2$ to $n = 3$, the H α line at $15,241.4 \text{ cm}^{-1}$
- 2) $n' = 3$ to $n = 4$, Infrared line at $5,334.4 \text{ cm}^{-1}$
- 3) $n' = 4$ to $n = 5$, Infrared line at $2,469.1 \text{ cm}^{-1}$.

For left circularly polarized probe radiation the

Selection rules are:

Any n , $\Delta l = \pm 1$, $\Delta m = 1$ - (1)

In note 306(5) it was shown that for the H α line there are five transitions possible when the selection rule (1) is applied. For the infra-red line at $5,334.4 \text{ cm}^{-1}$ there are ten transitions possible. (In note 306(5) there was a typo, eleven should have been ten). For the infra-red line at $2,469.1 \text{ cm}^{-1}$ there are seventeen possible transitions as shown below. So:

n'	Number of Transitions
2	5
3	10
4	17

So the number of possible transitions in left circular polarization ($\Delta m = 1$) is $n' - 2 + 1$.
 In linear polarization ($\Delta m = 0$), the number of possible transitions is $2n' - 1$.

2) For $h' = 4$ to $h = 5$ line the quantum numbers are as follows:

4s $n' = 4, \ell = 0, m = 0$

4p $k' = 4, l = 1, m = -1, 0, 1$

$$n' = 4, l = 2, m = -2, -1, 0, 1, 2$$

$$4f \quad n' = 4, \ell = 3, m = -3, -2, -1, 0, 1, 2, 3$$

$$\underline{5s} \quad n=5, \quad l=0, \quad m=0$$

$$5p \quad n = 5, \quad l = 1, \quad m = -1, 0, 1$$

$$n = 5, \quad l = 2, \quad m = -2, -1, 0, 1, 2$$

58 $n = 5, l = 3, m = -3, -2, -1, 0, 1, 2, 3$

$$Sg \quad n = 5, \ell = 4, \quad n = -4, -3, -2, -1, 0, 1, 2, 3, 4$$

Therefore for $\Delta n = 1$, the possible transitions are:

1) $4s \rightarrow 5p$ ($n'=4, l=0, m=0 \rightarrow n=5, l=1, m=1$).

2) $4p \rightarrow 5s$ ($n' = 4, l = 1, m = -1 \rightarrow n = 5, l = 0, m = 0$)

$$3 \left\{ \begin{array}{l} 3) \\ 4) \\ 5) \end{array} \right. 4p \rightarrow 5d \quad (\begin{matrix} l' = 4, \\ " \\ " \end{matrix}, \begin{matrix} l = 1, \\ 0 \\ 1 \end{matrix}, \begin{matrix} n = -1 \\ \\ 1 \end{matrix} \rightarrow \begin{matrix} n = 5, \\ " \\ " \end{matrix}, \begin{matrix} l = 2, \\ \\ 2 \end{matrix}, \begin{matrix} n = 0 \\ 1 \\ 2 \end{matrix})$$

$$\left[\begin{array}{l} 6 \\ 7 \end{array} \right] \quad 4d \rightarrow 5f \quad (n'=4, l=2, m=-2 \rightarrow n=5, l=3, m=-1)$$

7)	"	1	"	0
8)	"	0	"	1
9)	"	1	"	2
10)		2	"	3

$$4f \rightarrow 5g \quad (n' = 4, \ell = 3, m = -3 \rightarrow n = 5, \ell = 4, m = -2)$$

3) So the total number of lines is:

$$N = 2 + \sum_{l=1}^{n'-1} (2l+1) = n'^2 + 1 \quad - (2)$$

Q.E.D.

Therefore if left circularly polarized probe radiation is used, the far infra-red line at 81.52 cm^{-1} is split into 170 lines ($n'=13$ to $n=14$) as the result of basic quantum theory (the Planck distribution used is the Beer Lambert law). Therefore there are 170 Evans / Morris red shifts.

The microwave line at 1.704 cm^{-1} ($n'=50$ to $n=51$) is split into 2,501 lines. There are 2,501 Evans / Morris red shifts in the microwave.