The Lamb shift in atomic hydrogen calculated from Einstein-Cartan-Evans (ECE) field theory

M.W. Evans^{1*} and H. Eckardt^{2**}

*Alpha Institute for Advanced Studies (AIAS) (www.aias.us); **Unified Physics Institute of Technology (UPITEC) (www.upitec.org)

The Lamb shift in atomic H is calculated from the point of view of a generally covariant unified field theory - the Einstein Cartan Evans (ECE) field theory. The method adapted is to use an averaged vacuum potential that reproduces the g factor of the electron to experimental precision. This averaged potential is deduced from the Thomson radius of the photon and is used in the Schrödinger equation of atomic hydrogen (H) to increase the kinetic energy operator. The effect on the total energy levels of the H atom is then calculated and the Lamb shift deduced analytically without perturbation theory. It is found that the averaged vacuum energy affects each orbital of H in a different way through its effect on the electron proton distance. This effect is measured by distance r(vac) that is found by comparison with the experimental Lamb shift between the 2s and 2p orbitals of H. This is a generally applicable method that removes the problems of quantum electrodynamics.

Keywords: Einstein-Cartan-Evans (ECE) unified field theory, Lamb shift in atomic H.

1.1 Introduction

The Einstein-Cartan-Evans (ECE) unified field theory [1–12] is generally covariant in all its sectors and is a causal and objective field theory of physics. As such it is diametrically the opposite of quantum electrodynamics (QED), the method usually used to calculate the Lamb shift in atomic hydrogen (H) [13–15]. The QED method relies on hidden assumptions and adjustable parameters, so it is not a first principles theory. QED is a theory of special relativity which at the

¹e-mail: EMyrone@aol.com ²e-mail: horsteck@aol.com same time acausal and subjective, and violates special relativity, being therefore internally inconsistent on a basic level. The reason is that QED assumes that a point electron radiates a virtual photon that does not obey the Einstein equation of special relativity and cannot be observed experimentally. The virtual photon produces a virtual electron and a virtual positron, which again violate special relativity. These virtual electrons and positrons can go backwards or forwards in time, travel at any speed, and so on [13-15]. The point electron is affected by these virtual entities and is said to be shielded. These ideas emerged from the fact that QED is based on the Dirac equation of an electron and the equation of the quantized electromagnetic field, in which the potential is subject to second quantization and expressed in terms of creation and annihilation operators. A degree of precision is claimed for QED, but this is unjustifiable, because it is based on methods used to remove singularities in a perturbation method. These methods include mass re-normalization, dimensional regularization, and re-normalization of singularities introduced by perturbation theory; the arbitrary cut off of a series expansion, and the use of thousands of terms, usually put into diagrammatic form and known as Feynman diagrams. The result of the QED Lamb shift calculation is expressed in terms of the fine structure constant (α) . The latter's precision is however determined by its least precise constant, the Planck constant h. The experimentally measured relative standard uncertainty of h [16] is 1.7×10^{-7} and this is orders of magnitude greater than that of the electron g factor, which is 7.5×10^{-12} So QED can never produce the g factor to a precision greater than that of the Planck constant.

In Section 1.2, the ECE theory is used to calculate the Lamb shift in atomic H in the non-relativistic limit described by the Schrödinger equation. The average effect of the vacuum is deduced from the Thomson radius of the photon. This method has previously been used in ECE theory [1-12] to produce the g factor of the electron to the limit of precision of the fine structure constant (the precision determined by h as just argued). The g factor of the electron is increased in ECE theory as follows:

$$g \to g \left(1 + \frac{\alpha}{4\pi} \right)^2. \tag{1}$$

To calculate the Lamb shift with the Schrödinger equation, the kinetic energy operator is increased by the same amount as in Eq. (1). This vacuum effected kinetic energy operator, when used in the Schrödinger equation of the H atom, produces tiny changes in the electron to proton distance in each orbital of the H atom. The change is different for each orbital and is expressed in terms of the distance r(vac). The latter can be calculated analytically without perturbation theory. This method is first shown to produce the basic feature of the Lamb shift in atomic H, and is then refined analytically.

In Section 1.3 the results are graphed and used to illustrate the fact that this method is a generally applicable one, capable of being refined almost indefinitely by contemporary analytical and computational software.

1.2 A Simple non-relativistic calculation of the Lamb shift

The interaction of the electromagnetic field with the hydrogen atom is in ECE theory a problem of causal general relativity. In the non-relativistic quantum approximation to ECE theory [1-12] the electron and proton are bound together by a Coulomb potential and described by the well known öö equation of atomic H [17]:

$$\left(-\frac{\hbar^2}{2m}\nabla^2 - \frac{e^2}{4\pi\epsilon_0 r}\right)\psi = E\psi. \tag{2}$$

Here \hbar is the reduced Planck constant, m is the reduced mass, -e is the charge on the electron, e is the charge on the proton, e_0 is the vacuum permittivity and e is the proton to electron distance and radial coordinate of the spherical polar system of coordinates. The total energy levels \mathbf{E} of the \mathbf{H} atom are given by the Schrödinger equation as:

$$E = \frac{\alpha}{2n^2 a} \text{cm}^{-1} (\text{wavenumbers})$$
 (3)

where a is the Bohr radius:

$$a = \frac{4\pi\epsilon_0 \hbar^2}{m_e e^2} \tag{4}$$

and where α is the fine structure constant

$$\alpha = \frac{e^2}{4\pi\hbar c\epsilon_0}. (5)$$

Here n is the principle quantum number [17] and c is the speed of light in a vacuum. The Schrödinger equation produces the result:

$$E(2s) = E(2p) = \frac{\alpha}{8a} \text{cm}^{-1}$$
(6)

i.e. the energy levels of the 2s orbital and the three 2p orbitals are the same, i.e. degenerate. The Dirac equation of special relativistic quantum mechanics refines this result [18] to show that:

$$E\left(2s_{\frac{1}{2}}\right) = E\left(2p_{\frac{1}{2}}\right) \tag{7}$$

The observed Lamb shift is [17]:

$$E\left(2s_{\frac{1}{2}}\right) - E\left(2p_{\frac{1}{2}}\right) = 0.0353 \,\text{cm}^{-1} \tag{8}$$

and from Eqs. (6) and (8) produces a relative change:

$$\left(\frac{\Delta \overline{\mathbf{v}}}{\overline{\mathbf{v}}}\right)_{\text{exp.}} = 2.048 \times 10^{-7} \tag{9}$$

i.e. about two parts in ten million. The Lamb shift has no explanation in the Schrödinger and Dirac equations of the unperturbed H atom. The generally accepted explanation for the Lamb shift was first given by Bethe [19] as the effect of the quantized electromagnetic field's zeroth energy level. This is known somewhat misleadingly as the effect of the vacuum. More precisely it is due to the way in which the electromagnetic field is quantized in terms of harmonic oscillators. In order to approach this problem in ECE theory an electromagnetic potential is used to increase the value of the kinetic energy operator of Eq. (2). This method was first used [1–12] to calculate the g factor of the electron, and is based as follows on the Thomson radius of the photon.

The basic hypothesis is that:

$$\frac{\left\langle \epsilon A(\text{vac}) \right\rangle}{\hbar \kappa} = \frac{\alpha}{4\pi} \tag{10}$$

where A(vac) is the classical "vacuum potential", a term that is intended to denote the average effect of the electric and magnetic fields that exist in the absence of any photons. This idea comes from the fact that the lowest (or zero'th) energy level of the electromagnetic field quantized as a harmonic oscillator is

$$E = \left(n + \frac{1}{2}\right)\hbar\omega\tag{11}$$

where $\hbar\omega$ is the quantum of energy and n the number of photons. It is seen that there is an energy:

$$E_0 = \frac{1}{2}\hbar\omega \tag{12}$$

when there are no photons (n=0). This means [17] that there are fluctuating electric and magnetic fields present when there are no photons present. The linear momentum associated with these "vacuum fields" is denoted p(vac), and the linear momentum associated with $\hbar\omega$ is $\hbar\kappa$, where:

$$\kappa = \frac{\omega}{c} \tag{13}$$

is the wave-number magnitude. These ideas come from the fact that in quantum mechanics [17] the zero'th energy eigenvalue of the harmonic oscillator is non-zero.

The magnitude of A(vac) may be expressed in terms of the magnetic flux density magnitude $B^{(0)}$ associated with the quantum of energy $\hbar\omega$ [1–12]. Thus

$$B^{(0)} = \frac{\mu_0 ce}{Ar} \tag{14}$$

where μ_0 is the vacuum permeability defined by:

$$\epsilon_0 \mu_0 = \frac{1}{c^2} \tag{15}$$

and where Ar is the surface area associated with the energy quantum $\hbar\omega$. This is obtained from the Thomson radius:

$$r = \frac{1}{\kappa} \tag{16}$$

and so the surface area of the sphere with this radius is:

$$Ar = 4\pi r^2 = \frac{4\pi}{\kappa^2}. ag{17}$$

Thus:

$$B^{(0)} = \kappa |A(\text{vac})| = \kappa A^{(0)} = \frac{\mu_0 c e \kappa^2}{4\pi}$$
 (18)

and:

$$eA^{(0)} = \frac{e^2\kappa}{4\pi\epsilon_0 c}. (19)$$

The fine structure constant is therefore defined by:

$$\alpha = \frac{eA^{(0)}}{\hbar\kappa} = \frac{e^2}{4\pi\epsilon_0\hbar c}.$$
 (20)

The average value of $A^{(0)}$ is now found by averaging over the polar angles: recall that the surface area of a sphere of radius r is:

$$S = \int_0^{2\pi} d\phi \int_0^{\pi} r^2 \sin\theta d\theta = 4\pi r^2$$
 (21)

so for unit radius:

$$\int_0^{2\pi} d\phi \int_0^{\pi} \sin\theta d\theta = 4\pi. \tag{22}$$

Thus:

$$\left\langle A^{(0)} \right\rangle = \int_0^{2\pi} \int_0^{\pi} A^{(0)} \left(\theta, \phi \right) d\phi \sin \theta d\theta = 4\pi A^{(0)}. \tag{23}$$

This procedure means that the quantum of momentum $\hbar \kappa$ with Thomson radius $1/\kappa$ is averaged over θ and ϕ of the spherical polar coordinate system, i.e. $A^{(0)}$ is averaged over all possible orientations. The result is Eq. (10), which comes from a quantized electromagnetic field.

Another method of justifying the factor 4π is to consider the magnetic flux associated with the quantum of electromagnetic energy $\hbar\omega$ [1–12]:

$$\Phi^{(0)} := \frac{\mu_0 ce}{4\pi} = \frac{e}{4\pi\epsilon_0 c} \tag{24}$$

it is known [13] that the quantum of magnetic flux can also be expressed as:

$$\Phi = \frac{\hbar}{e} \tag{25}$$

so:

$$\Phi^{(0)} = \alpha \Phi. \tag{26}$$

In fact:

$$\frac{e}{4\pi\epsilon_0 c} = \alpha \frac{\hbar}{e} \tag{27}$$

SO

$$\Phi^{(0)} = \alpha \Phi \tag{28}$$

Q. E. D.

From Eq. (10) we may express $\langle A^{(0)} \rangle$ as follows:

$$\left\langle A^{(0)} \right\rangle = \frac{e}{16\pi^2 \epsilon_0 \hbar c^2} \left\langle E^{(0)} \right\rangle \tag{29}$$

where $E^{(0)}$ is the zero'th point energy associated with the quantized electromagnetic field. Now extend Eq. (29) to energy momentum:

$$\langle eA_{\mu}(\text{vac})\rangle \frac{\alpha}{4\pi} \langle p_{\mu}(\text{vac})\rangle.$$
 (30)

It is seen that this is a type of minimal prescription [13–15] that accounts for the vacuum energy. More accurately, vacuum energy should be termed space-time energy. The effect of $\langle A_{\mu}(vac) \rangle$ is as follows:

$$p_{\mu} \to \left(1 + \frac{\alpha}{4\pi}\right) p_{\mu}. \tag{31}$$

So the Dirac equation of the electron changes to [1-12]:

$$\left(\gamma^{\mu} p_{\mu} \left(1 + \alpha'\right) - mc\right) \psi = 0,
\alpha' := \frac{\alpha}{4\pi}.$$
(32)

This is equivalent to an increase in the Dirac matrix:

$$\gamma^{\mu} \to (1 + \alpha')\gamma^{\mu}. \tag{33}$$

In QED this increase is denoted:

$$\gamma^{\mu} \to \left(1 + \Lambda^{(2)}\right) \gamma^{\mu} \tag{34}$$

where $\Lambda^{(2)}$ is a convergent vertex [13-15], a kind of adjustable parameter.

The well known factor 2 of the Dirac theory of the electron interacting with a classical electromagnetic field originates geometrically in the relation between the Minkowski metric $\eta^{\mu\nu}$ and the Dirac matrices [13]:

$$2\eta^{\mu\nu} = \gamma^{\mu}\gamma^{\nu} + \gamma^{\nu}\gamma^{\mu}. \tag{35}$$

Both Dirac matrices change according to Eq. (33), i.e.:

$$\gamma^{\mu} \rightarrow (1 + \alpha')\gamma^{\mu},$$
 (36)

$$\gamma^{\mathsf{v}} \to (1 + \alpha')\gamma^{\mathsf{v}},$$
 (37)

so:

$$\gamma^{\mu}\gamma^{\nu} + \gamma^{\nu}\gamma^{\mu} \rightarrow 2(1 + \alpha')^{2}(\gamma^{\mu}\gamma^{\nu} + \gamma^{\nu}\gamma^{\mu}) \tag{38}$$

and the g factor of the electron, g = 2, changes to:

$$g(ECE) = 2\left(1 + \frac{\alpha}{4\pi}\right)^2. \tag{39}$$

Note that this result happens to agree exactly to first order in a with the well known calculation of Schwinger [13] in QED:

$$g\left(\text{Schwinger}\right) = 2 + \frac{\alpha}{\pi} \tag{40}$$

and gives a small correction to second order in α .

The Schrödinger equation (2) is changed therefore by the same factor, appearing in the kinetic energy operator. The latter is defined by the classical:

$$T = \frac{p^2}{2m} \tag{41}$$

and the operator equation:

$$p \to -i\hbar \nabla$$
. (42)

So the kinetic energy operator is:

$$\hat{T} = -\frac{\hbar^2 \nabla^2}{2m}.\tag{43}$$

This appears as the square of momentum, so:

$$\nabla^2 \to \left(1 + \frac{\alpha}{4\pi}\right)^2 \nabla^2. \tag{44}$$

Our method of calculating the Lamb shift is therefore based on the same method as used to calculate g of the electron [1–12]. Before proceeding, we discuss the precision of the ECE method for g of the electron.

From ECE theory:

$$g = 2\left(1 + \frac{\alpha}{4\pi}\right)^2 = 2 + \frac{\alpha}{\pi} + \frac{\alpha^2}{8\pi^2}$$
 (45)

with basic definition:

$$\alpha := \frac{e^2}{4\pi\epsilon_0 \hbar c}.\tag{45a}$$

The fundamental constants of physics are agreed upon by treaty in standards

laboratories such as www.nist.gov of the US National Institute for Standards and Technology. This site gives:

$$h(\text{exptl.}) = (6.6260693 \pm 0.0000011) \times 10^{-34} J_s$$
 (47)

$$e(\text{exptl.}) = (1.60217653 \pm 0.0000000 \, 14) \times 10^{-19} C$$
 (48)

$$c(\text{exact}) = 2.99792458 \times 10^8 \text{ms}^{-1} \tag{49}$$

$$\epsilon_0(\text{exact}) = 8.854187817 \times 10^{-12} \text{J}^{-1} \text{C}^2 \text{m}^{-1}$$
 (50)

$$\mu_0(\text{exact}) = 4\pi \times 10^{-7} J s^2 C^{-2} m^{-1}$$
 (51)

with relative standard uncertainties. By treaty:

$$\mu_0 \epsilon_0 = \frac{1}{c^2} \tag{52}$$

is taken as exact. With a sufficiently precise value of π :

$$\pi = 3.141592653590.$$
 (53)

Using these values in Eq. (45a) gives:

$$\alpha = 0.007297(34) \tag{54}$$

where the result has been rounded off to the relative standard uncertainty of h, the least precisely known constant of α . Thus α cannot be more precisely known than h. The latter is determined by an experimental method described on www. nist.gov. Now use Eq. (54) in Eq. (45) to give the theoretical value of g from ECE theory:

$$g(ECE) = 2.002323(49).$$
 (55)

The experimental value of g is known to much greater precision than h, and is:

$$g(\text{exptl.}) = 2.0023193043718 \pm 0.000000000075.$$
 (56)

It is seen that the difference is:

$$g(ECE) - g(exptl.) = 0.000004$$
 (57)

which is about the same order of magnitude as the experimental uncertainty in h. It is concluded that ECE gives g as precisely as the uncertainty of h will allow. The latter is not given consistently in the literature. For example, a much used text such as Atkins [17] gives it as:

$$h(\text{Atkins}) = 6.62618 \times 10^{-34} J_{s}$$
 (58)

without uncertainty estimates. This is very different (fourth decimal place) from the NIST value. Despite this, Atkins [17] gives:

$$g(Atkins) = 2.002319314 \tag{59}$$

which is different from Eq. (56) only in the eighth decimal place. This is severely self-inconsistent. Atkins gives the g factor of the electron as:

$$g(Atkins) = 2.002319314 \tag{60}$$

which is different from the NIST value in the eighth decimal place, while it is claimed at NIST that $g(\exp)$ from Eq. (56) is accurate to the twelfth decimal place. This is a another major inconsistency between well used current sources. Ryder [13] gives:

$$g(\text{Ryder}) = 2.0023193048$$
 (61)

which is different from the NIST value in the tenth decimal place, and Ryder claims that QED gives g to this same precision. As we have seen, this cannot be true because the uncertainty of any theoretically calculated g is limited by the uncertainty on h. In summary:

$$g(Schwinger) = 2 + \frac{\alpha}{\pi} = 2.002322(8)$$
 (62)

$$g(ECE) = 2 + \frac{\alpha}{\pi} + \frac{\alpha^2}{8\pi^2} = 2.002323(49)$$
 (63)

$$g(NIST) = 2.0023193043718 \pm 0.00000000000075 \tag{64}$$

$$g(Atkins) = 2.002319314 \pm (?)$$
 (65)

$$g(Ryder) = 2.0023193048 \pm (?)$$
 (66)

and there is little doubt that other textbooks and sources will give further different values of g. So where does this finding leave the claims of QED? The Wolfram site claims that QED gives g using the series:

$$g = 2\left(1 + \frac{\alpha}{2\pi} - 0.328\left(\frac{\alpha}{\pi}\right)^2 + 1.181\left(\frac{\alpha}{\pi}\right)^3 - 1.510\left(\frac{\alpha}{\pi}\right)^4 + \dots + 4.393 \times 10^{-12}\right)$$
(67)

which is derived from thousands of Feynman diagrams (sic). However, the various terms in Eq. (67) come from all the assumptions listed in the introduction, and so this cannot be a first principles theory despite its elaborate nature.

An even worse internal self-inconsistency emerges because at NIST the fine structure constant is claimed to be:

$$\alpha(\text{NIST}) = (7.297352560 \pm 0.000000024) \times 10^{-3}$$
(68)

both experimentally and theoretically. The first thing to note is that this claim, Eq. (68), is different in the eighth decimal place from Eq. (63), which is calculated from NIST's own fundamental constants, Eqs. (55) to (60). So the NIST website is internally inconsistent to an alarming degree, because it is at the same time claimed that Eq. (68) is accurate to the tenth decimal place. From Eq. (56) it is seen that h at NIST is accurate only to the sixth decimal place, which limits the accuracy of to this, i.e. ten thousand times less precise than claimed. The source of this very large internal discrepancy must therefore be found. The theoretical claim for the fine structure constant at NIST comes from QED, which is described as a theory in which an electron emits a virtual photon, which emits virtual electron positron pairs. The virtual positron is attracted and the virtual electron is repelled from the real electron. This process results in a screened charge, and the fine structure constant in QED is the square of an entity known as the completely screened charge, a mathematical limiting value. This is defined as the limit of zero momentum transfer or in the limit of infinite distance. At high energies the fine structure constant drops to 1/128, and so is not a constant at all. It certainly cannot be claimed to be precise to the relative standard uncertainty of Eq. (68), taken directly from the NIST website itself (www.nist.gov).

The experimental claim for the fine structure constant at NIST comes from the use of the quantum Hall effect combined with a calculable cross capacitor to measure the standard resistance. The immediate problem with this method is the use of the von Klitzing constant:

$$R_{\kappa} = \frac{h}{e^2} = \frac{\mu_0 c}{2} (\text{sic}) \tag{69}$$

which is clearly limited once more to the accuracy of h. The latter is found at NIST by a different experimental method, giving Eq. (56). The accuracy of the charge on the proton is only ten times better than h, so R_K cannot be more accurate than h. If α were really as accurate as claimed in Eq. (68), then R and $\frac{h}{e^2}$ would also be known to this accuracy, conflicting with Eqs. (56) and (57)

In view of these serious problems it is considered that the claim of QED to reproduce the Lamb shift to high accuracy is also in complete doubt, as are all the so-called "precision tests" of QED, for example the g of the electron, the Lamb shift, positronium and so forth. The only sensible way forward is to calculate the Lamb shift in terms of r(vac) and to find r(vac) from the experimental value of the Lamb shift, making no unjustifiable claims of precision. This aim is achieved as follows on the non-relativistic level and can be considerably refined in future work.

The distance r(vac) is defined from Eq. (2) and (44) by:

of the same NIST website.

$$-\frac{\hbar^2 \nabla^2}{2m} \left(\frac{\alpha}{2\pi} + \frac{\alpha^2}{16\pi^2} \right) \psi = \frac{e^2}{4\pi\epsilon_0} \left(\frac{1}{r} - \frac{1}{r + r(\text{vac})} \right) \psi \tag{70}$$

i.e. the effect of the vacuum potential is considered to be a shift in the electron to proton distance for each orbital. To first order in α :

$$\nabla^2 \psi = -\left(\frac{4mc\pi}{\hbar}\right) \left(\frac{1}{r} - \frac{1}{r + r(\text{vac})}\right) \psi \tag{71}$$

Using the normalization given by Atkins [17] we consider five of the H orbitals:

$$\psi(1s) = \frac{1}{\pi^{\frac{1}{2}}} \left(\frac{1}{a}\right)^{\frac{3}{2}} \exp\left(-\frac{r}{a}\right),\tag{72}$$

$$\psi(2s) = \frac{1}{4\sqrt{2}\pi^{\frac{1}{2}}} \left(2 - \frac{r}{a}\right) \exp\left(-\frac{r}{2a}\right),\tag{73}$$

$$\Psi(2p_z) = \frac{1}{2} \left(\frac{3}{\pi}\right)^{\frac{1}{2}} \cos\theta \left(\frac{1}{a}\right)^{\frac{3}{2}} \left(\frac{1}{2\sqrt{6}}\right) \frac{r}{a} \exp\left(-\frac{r}{2a}\right),\tag{74}$$

$$\Psi(2p_X) = -\frac{1}{2} \left(\frac{3}{2\pi}\right)^{\frac{1}{2}} \sin\theta e^{i\phi} \left(\frac{1}{a}\right)^{\frac{3}{2}} \left(\frac{1}{2\sqrt{6}}\right) \frac{r}{a} \exp\left(-\frac{r}{2a}\right),\tag{75}$$

$$\Psi(2p_Y) = -\frac{1}{2} \left(\frac{3}{2\pi}\right)^{\frac{1}{2}} \sin\theta e^{-i\phi} \left(\frac{1}{a}\right)^{\frac{3}{2}} \left(\frac{1}{2\sqrt{6}}\right) \frac{r}{a} \exp\left(-\frac{r}{2a}\right),\tag{76}$$

In spherical polar coordinates the laplacian operator is [17]:

$$\nabla^2 \psi = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial \psi}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \psi}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \psi}{\partial \phi^2}$$
 (77)

Computer algebra may now be used to evaluate r(vac) for each orbital. We first consider for the sake of illustration a special case of the $2p_z$ orbital, the maximum:

$$\cos \theta = 1. \tag{77a}$$

It is found that:

$$\frac{r(\text{vac})}{r(r+r(\text{vac}))}(1s) = \frac{1}{2\pi} \cdot \frac{\hbar}{mca} \left(\frac{1}{r} - \frac{1}{2a}\right),\tag{78}$$

$$\frac{r(\text{vac})}{r(r+r(\text{vac}))}(2s) = \frac{1}{2\pi} \cdot \frac{\hbar}{mca} \left(\frac{1}{r} - \frac{1}{8a}\right),\tag{79}$$

$$\frac{r(\text{vac})}{r(r+r(\text{vac}))} \left(2p_{Z_s}\cos\theta = 1\right) = \frac{1}{2\pi} \cdot \frac{\hbar}{mca} \left(\frac{1}{r} - \frac{1}{8a} - \frac{a}{r^2}\right). \tag{80}$$

This checks that the method gives the correct qualitative result. A remarkable simplification occurs to show that the Lamb shift is proportional to:

$$\frac{r(\operatorname{vac})(2s)}{r(r+r(\operatorname{vac}))} - \frac{r(\operatorname{vac})(2p_Z,\cos\theta=1)}{r(r+r(\operatorname{vac}))} = \frac{1}{4\pi} \cdot \frac{\hbar}{mc} \cdot \frac{1}{r^2}$$
(15 81)

More generally the Lamb shift will depend on the polar coordinates. Despite its huge complexity, QED is unable to give these detailed results, which are

found by dispensing with perturbation theory and virtual particles, and calculating analytically. Finally the constant of proportionality in Eq. (72) has to be found. The potential energy of the H atom in wave-numbers is:

$$V_0 = -\frac{\alpha}{r} \tag{82}$$

before the vacuum perturbation and is:

$$V = -\frac{\alpha}{r + r(\text{vac})} \tag{83}$$

after the vacuum perturbation. So the change in potential energy is positive:

$$\Delta V = |V - V_0| = \alpha \left(\frac{1}{r} - \frac{1}{r + r(\text{vac})}\right) = \alpha \left(\frac{r(\text{vac})}{r(r + r(\text{vac}))}\right)$$
(84)

Note that the factor in Eq. (86) is obtained from the basic assumption used in this paper, that the Schrödinger equation of H with radiative correction, to first order in a:

$$-\frac{\hbar^2}{2m}\left(1 + \frac{\alpha}{2\pi}\right)\nabla^2\psi - \frac{e^2}{4\pi\epsilon_0 r}\psi = E\psi \tag{85}$$

is equivalent to:

$$-\frac{\hbar^2}{2m}\nabla^2\psi - \frac{e^2}{4\pi\epsilon_0(r+r(\text{vac}))}\psi = E\psi. \tag{86}$$

It is assumed that r(vac) is small so the wave-function in Eq. (86) is to a very good approximation the wave-function of H in the absence of the radiative correction.

$$\Psi = \Psi_0. \tag{87}$$

Therefore it follows that:

$$\nabla^2 \Psi_0 = -4\pi \frac{mc}{\hbar} \left(\frac{1}{r} - \frac{1}{r + r(\text{vac})} \right) \Psi_0$$
 (88)

and from the above calculation we obtain:

$$\frac{1}{r + r_{2p} \left(\text{vac}\right)} - \frac{1}{r + r_{2s} \left(\text{vac}\right)} = \frac{1}{2\pi^{\frac{3}{2}}} \frac{\hbar}{mc} \frac{1}{r^2}.$$
 (89)

The change in potential energy is therefore:

$$\Delta V = \frac{\alpha}{2\pi^{\frac{3}{2}}} \frac{\hbar}{mc} \frac{1}{r^2} \tag{90}$$

and the change in total energy is:

$$\Delta E = \frac{r}{2n^2 a} \Delta V = \left(\frac{1}{16\pi^{\frac{3}{2}}} \frac{\alpha}{a} \frac{\hbar}{mc}\right) \frac{1}{r} = 0.0353 \,\text{cm}^{-1}$$
 (91)

which is the Lamb shift in atomic H. Here r is:

$$r = 1.69 \times 10^{-7} \text{ m.}$$
 (92)

From Eq. (89):

$$\frac{r_{2s}(\text{vac}) - r_{2p}(\text{vac})}{(r + r_{2p}(\text{vac}))(r + r_{2s}(\text{vac}))} = \frac{1}{2\pi^{\frac{3}{2}}} \frac{\hbar}{mc} \frac{1}{r^2}.$$
(93)

Eq. (89) implies that:

$$r \gg r_{2s} \left(\text{vac} \right) \sim r_{2p} \left(\text{vac} \right) \tag{94}$$

so in this approximation, Eq. (93) becomes:

$$r_{2s}(\text{vac}) - r_{2p}(\text{vac}) \sim \frac{1}{2\pi^{\frac{3}{2}}} \frac{\hbar}{mc}$$
 (95)

i.e.

$$r_{2s}\left(\operatorname{vac}\right) - r_{2p}\left(\operatorname{vac}\right) \sim \frac{1}{2\pi^{\frac{5}{2}}} \frac{h}{mc}$$
(96)

where the standard Compton wavelength is:

$$\frac{h}{mc} = 2.426 \times 10^{-12} \,\mathrm{m}.\tag{97}$$

Thus we arrive at:

$$r_{2s}(\text{vac}) - r_{2p}(\text{vac}) \sim 3.48 \times 10^{-13} \,\text{m}.$$
 (98)

This is a sensible result because the classical electron radius is:

$$r(\text{classical}) = \frac{1}{4\pi\epsilon_0} \frac{e^2}{mc^2} = 2.818 \times 10^{-15} \,\text{m}$$
 (99)

and the Bohr radius is:

$$a = 5.292 \times 10^{-11} \text{ m}$$
 (100)

so the radiative correction perturbs the electron orbitals by about ten times the classical radius of the electron, and by orders less than the Bohr radius. This is a remarkable result because it shows why the Lamb shift is constant as observed. For a given orientation defined by:

$$\cos \theta = 1 \tag{101}$$

the shift is determined completely by $\frac{1}{r}$ within a constant of proportionality

$$r = \frac{1}{32\pi^{\frac{3}{2}}} \frac{\alpha}{a} \frac{\hbar}{mc}.$$
 (102)

The next section discusses the angular dependence of the Lamb shift and gives some graphical results. Finally in this section, consideration is given to an averaged 2p orbital defined by:

$$\langle \psi(2p) \rangle = \left(\psi^2(2p_Z) - \psi(2p_X) \psi(2p_Y) \right)^{\frac{1}{2}}$$

$$= 2 \left(1 - \frac{r}{a} \right) \exp\left(-\frac{r}{2a} \right)$$
(103)

within a normalization factor. The Lamb shift is then evaluated from:

$$\nabla^2 \psi(2s) = -4\pi \frac{mc}{\hbar} x_1 \psi(2s) \tag{104}$$

and

$$\nabla^{2} \langle \psi(2p) \rangle = -4\pi \frac{mc}{\hbar} x_{2} \langle \psi(2p) \rangle \tag{105}$$

so that

$$x_1 = \frac{\hbar}{\pi mc} \cdot \frac{1}{2a} \left(\frac{1}{r} - \frac{1}{8a} \right) \tag{106}$$

$$x_2 = \frac{\hbar}{\pi mc} \cdot \frac{1}{2a} \left(\frac{1}{r} - \frac{a}{r^2} - \frac{1}{8a} \right). \tag{107}$$

The difference in these values gives the Lamb shift between the 2s state in Hand the 2p state averaged as in Eq. (94), with all angular dependence removed. This again gives Eq. (81).

1.3 Angular dependence of the Lamb shift and graphical results

In the following we present two further models for calculating the Lamb shift with averaged 2p orbitals of H. According to Eq. (71) the Lamb shift follows from

$$\nabla^2 \psi = -\frac{4\pi mc}{\hbar} x \psi \tag{108}$$

with

$$x = \frac{1}{r} - \frac{1}{r + r(\text{vac})}.$$
 (109)

The natural way would be to use the full angular dependence of the wavefunctions (Eqs. 72–76) in these equations. However, applying the full laplacian operator (77) to the 2p orbitals and inserting the results into (108) gives the same result as for the 2s orbital, namely Eq. (79). The detailed computer algebra calculation shows that the θ and Φ dependencies after application of the laplacian are the same as for the wavefunctions itself. So the angular dependence cancels out in Eq. (108). The remaining factors lead to Eq. (79) again. In the following we inspect two alternative methods of calculating the Lamb shift by averaged 2p orbitals.

First method

We use the 2p mean wavefunction defined by

$$\langle \psi(2p) \rangle := \left(\psi^2(2p_Z) - \psi(2p_X) \psi(2_{p_Y}) \right)^{\frac{1}{2}}$$
(110)

The difference wave function responsible for the experimentally observed Lamb shift is then given by

$$\psi := \psi(2s) - \langle \psi(2_p) \rangle. \tag{111}$$

By inserting (72)–(76) this leads to

$$\psi = 2\left(1 - \frac{r}{a}\right) \exp\left(-\frac{r}{2a}\right) \tag{112}$$

The normalization factors were omitted because they cancel out in (108). We see that the angular dependence has disappeared, no further averaging is needed. Evaluation of (108) gives

$$x = \frac{\hbar}{\text{mc}\pi} \frac{1}{16a^2} \frac{-r^2 + gar - 12a^2}{r(r-a)}$$
 (113)

With

$$\nabla V = -\alpha x \tag{15 114}$$

we obtain

$$\Delta E = -\frac{\alpha x r}{2n^2 \alpha} = \frac{\hbar \alpha}{\text{mc} \pi} \frac{1}{128a^3} \frac{r^2 - 9ar + 12a^2}{r - a}$$
 (115)

Setting ΔE to the experimental value of 0.0353 cm⁻¹ (or 4.372 · 10⁻⁶ eV = 0.1607·10⁻⁶ Hartree) yields the effective radius values (in Bohr radius a_B):

$$r_1 = 1.63 \ a_B$$
 (116a)

$$r_2 = 7.37 \ a_{_{R}}$$
 (116b)

Compared to the result (92) this is very small, the Lamb shift is highly overestimated in this approximation. The reason probably is that Eq. (108) is set up for the difference wave function ψ . Although (108) is linear in ψ , it is not in handling the vacuum effect radius r(vac) as can be seen from Eq. (109). So we work out an alternative method.

Second method

The second method uses separate expressions of the effects of r(vac) for each orbital $\psi(2s)$ and $\langle \psi(2p) \rangle$ in correspondence to Eqs. (78–80):

$$\nabla^2 \psi(2s) = -\frac{4mc\pi}{\hbar} x_1 \psi(2s) \tag{117}$$

$$\nabla^2 \langle \psi(2p) \rangle = \frac{4mc\pi}{\hbar} x_2 \langle \psi(2p) \rangle \tag{118}$$

with averaged orbitals $\langle \psi(2p) \rangle$ asd defined by (110). For Eq. (116) we can directly take over the result of Eq. (79):

$$x_{1} = \frac{\hbar}{\text{me}\pi} \frac{1}{2a} \left(\frac{1}{r} - \frac{1}{8a} \right). \tag{119}$$

Evaluation of (118) leads to

$$x_2 = \frac{\hbar}{\text{me}\pi} \frac{1}{2a} \left(\frac{1}{r} - \frac{a}{r^2} - \frac{1}{8a} \right). \tag{120}$$

The difference term for the 2s-2p Lamb shift then is

$$x_1 - x_2 = \frac{\hbar}{\pi \text{mc}} \frac{1}{2r^2} \tag{121}$$

This is exactly result (81) which was obtained for $\psi(2pz)$ with $\cos\theta = 1$. Therefore the Lamb shift for this angular averaged model is the same as for the $2p_z$ orbital in the direction of maximum amplitude.

Now we discuss the radial dependence of some characteristic terms in graphical representation. All figures are in atomic units. The kinetic energy distribution

$$-\frac{\hbar^2}{2m}\nabla^2\psi(1s)(r)\cdot 4\pi r^2\tag{122}$$

of the 1s orbital is shown in Fig. 1a. There is a zero transition for r = 2. The corresponding transition also appears in the inverse radius difference (109). This can be seen from Fig. 1b where this difference is plotted, showing the effective potential of the radiative effects. The r dependence of the vacuum interaction radius r(vac) itself can be calculated from Eq. (78):

$$r(\text{vac})(1s) = \frac{2ar - r^2}{r + 4\beta a^2 - 2a}$$
 (123)

with

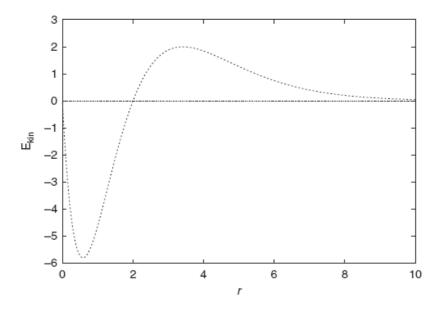


Fig. 1a. Kinetic energy radial distribution function of H 1s orbital.

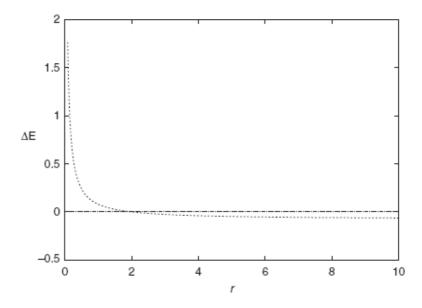


Fig. 1b. Difference term of potential energy for H 1s orbital.

$$\beta = \frac{\pi mc}{\hbar} \tag{124}$$

This function is graphed in Fig. 1.1c. r(vac) is positive for $r \le 2$ and negative above. The zero crossing corresponds to Figs. 1.1a and 1.1b. A negative radius r(vac) means that there is a decrease in potential instead of increase due to vacuum effects.

The same three graphs are shown for the 2s orbital in Figs. 1.2a-c. Here only the zero in kinetic energy at r=8 is also occurring in the effective potential. r(vac) is a second degree rational function of r, similar to the 1s case:

$$r(\text{vac})(2s) = \frac{8ar - r^2}{r + 1b\beta a^2 - 8a}$$
 (125)

Its sign coincides with that of the effective potential of Fig. 2b.

Corresponding results apply to the $2p_z$, orbital (Figs. 3a-c). Here both zeros in kinetic energy are visible in the effective potential of r(vac). Compared to the 2s orbital, the vacuum interaction radius itself has an analytical dependence of one degree higher in r:

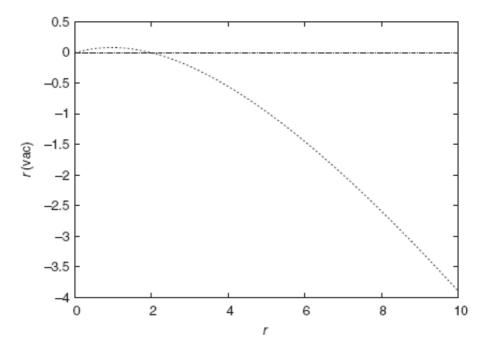


Fig. 1c. Vacuum interaction radius r(vac) for H 1s orbital.

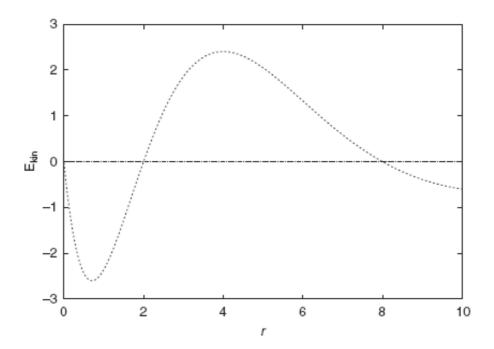


Fig. 2a. Kinetic energy radial distribution function of H 2s orbital.

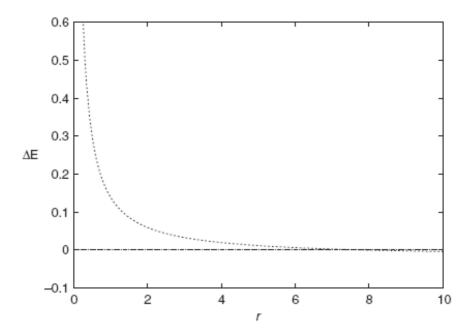


Fig. 2b. Difference term of potential energy for H 2s orbital.

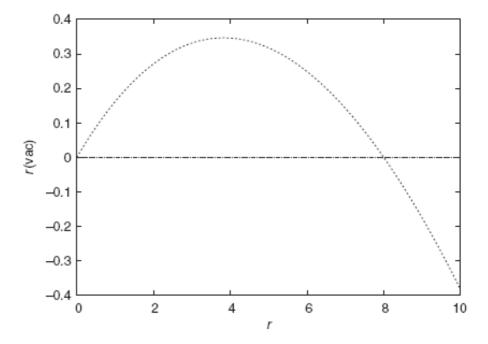


Fig. 2c. Vacuum interaction radius r(vac) for H 2s orbital.

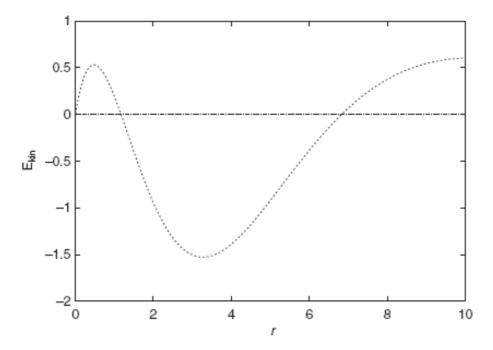


Fig. 3a. Kinetic energy radial distribution function of H 2pZ orbital.

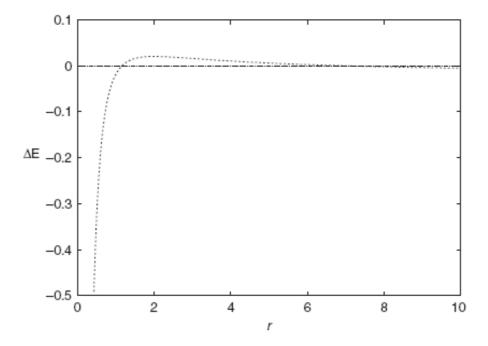


Fig. 3b. Difference term of potential energy for H 2pZ orbital.

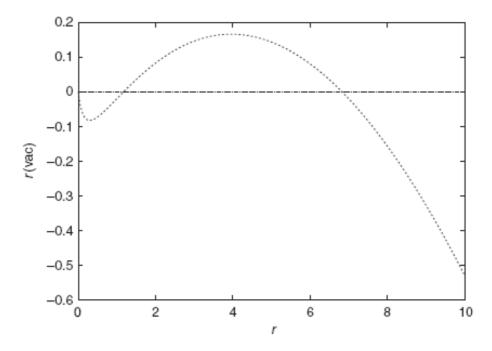


Fig. 3c. Vacuum interaction radius r(vac) for H $2p_z$ orbital.

$$r(\text{vac})(2p_z) = \frac{-r^3 + 8a^2r}{r^2 + (16\beta a^2 - 8a)r + 8a^2}$$
 (126)

All r(vac) functions of the three orbitals have in common that they are positive in regions where the wave function has significant values which can be seen from the kinetic energy distribution diagrams. This means that effectively energy is transferred from the quantum background to the atom. In the outer regions there is a small energy transport in the opposite direction.

Finally we present the difference terms between 2 s and 2p which make up the experimental Lamb shift energy difference. Fig. 4 shows the difference of the potential energy terms (Figs. 2b, 3b). There is a positive energy contribution near to the core which effects the positive sign of the Lamb shift energy. According to Eq. (93) it has a simple hyperbolic form. The difference of the radii itself is graphed in Fig. 5. The absolute values are small compared to the atomic radius, representing the fact that the Lamb shift is a relative energy correction of the order 10^{-7} .

We conclude that in this paper all important aspects of the Lamb shift could be made plausible on base of ECE theory. Calculations were carried out in the non-relativistic limit by using the Schrödinger equation of atomic Hydrogen. Angular dependent effects play no significant role.

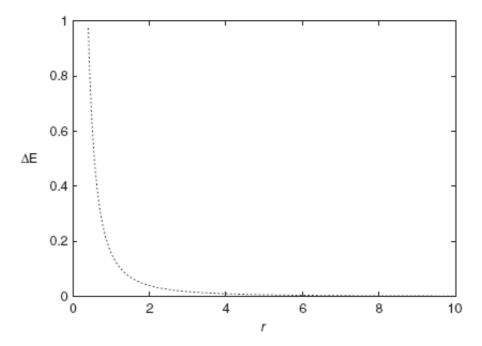


Fig. 4. Difference of potential terms for Lamb shift 2s-2pZ.

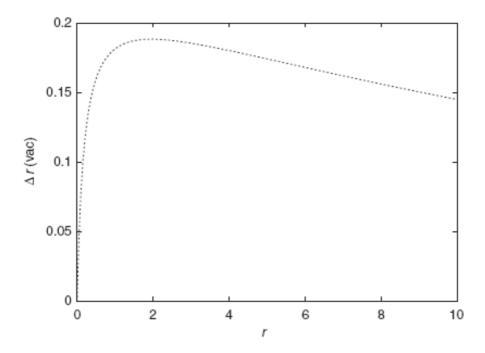


Fig. 5. Effective Lamb shift radius r(vac)(2s) - r(vac)(2pZ).

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